

Measuring N₂O and CH₄ exchange between the soil and atmosphere

Stephen Livesley

Department of Forest and Ecosystem Science

Key references:

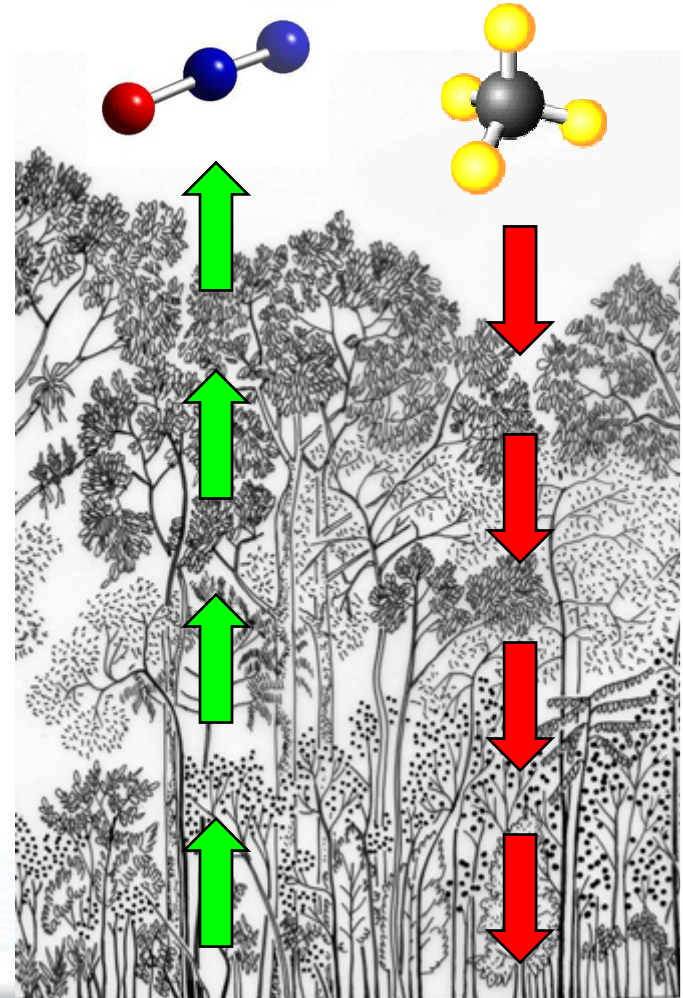
Matson & Harriss (1995) Biogenic trace gases: measuring emissions from soil & water

Denmead (2008) Plant and Soil 309

Parkin et al. (2003) <http://gracenet.usda.gov/GRACEnetTraceGasProtocol.pdf>

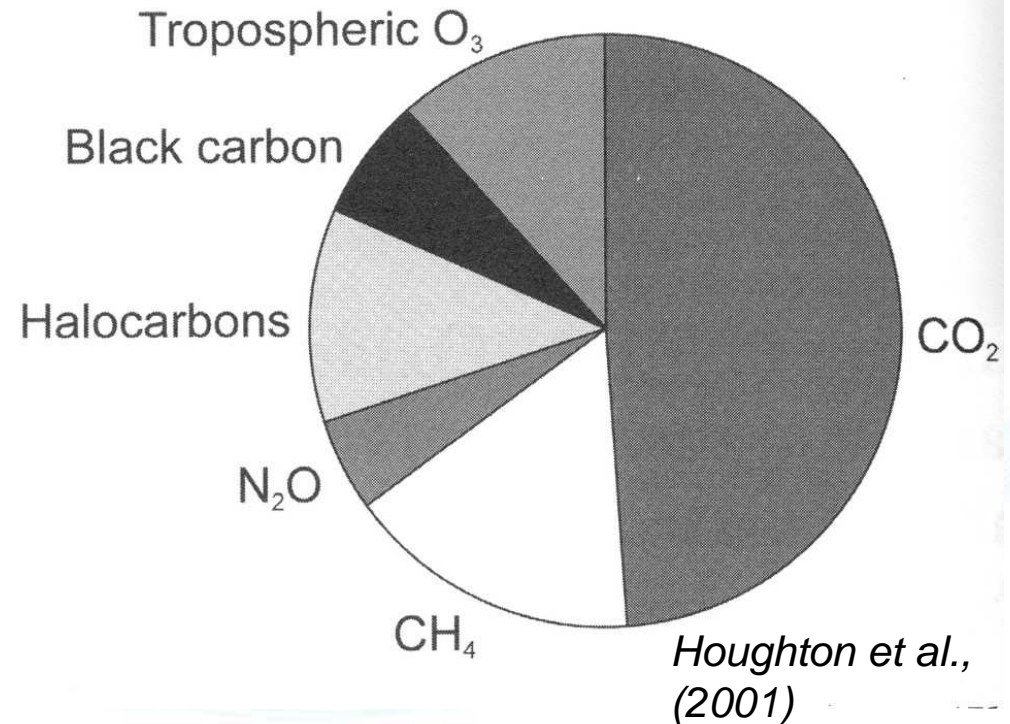
Soils & non-CO₂ exchange

- N₂O and CH₄ , GWP and CO₂-e
- Processes controlling N₂O & CH₄ flux
- Ways to measure N₂O and CH₄ flux
- Chamber measurement approaches
- Chamber measurement issues
- Examples of N₂O & CH₄ flux measures
- Conclusions



Global C cycle and non-CO₂

- Non-CO₂ gases rarely included in global C budgets,
- Sources and sinks not well understood.
- **CO₂** represents < 50% of current atmospheric radiative forcing.
- **Methane (CH₄)** and **Nitrous oxide (N₂O)** represent ~25% of radiative forcing.



Global C cycle and non-CO₂

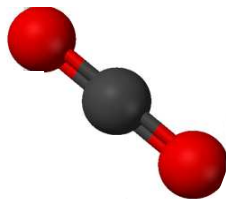
- **Global warming potential (GWP)** relates all GHG's to the radiative forcing of CO₂, based on absorption of radiation and persistence in atmosphere.

	Pre-ind.	Current	<i>GWP</i>
CO ₂	280	375	1
CH ₄	0.80	1.78	21
N ₂ O	0.28	0.31	310

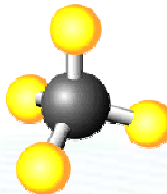
- **Carbon dioxide equivalents (CO₂-e)**, normalise all gases to that of CO₂ using their GWP.

Global C cycle and non-CO₂

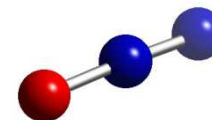
- Both CH₄ and N₂O sink-source mechanisms are vulnerable to human interference and feedback responses.
- A **multi-gas approach** is necessary. (*Robertson et al., 2000*)
e.g - N fertiliser may increase crop NPP and C sink. (*Crutzen 2008*)
BUT this may be offset by concurrent N₂O emissions.
- An **Imbedded** or **System** CO₂-e cost should be considered,
e.g. the C cost of fertiliser manufacture.



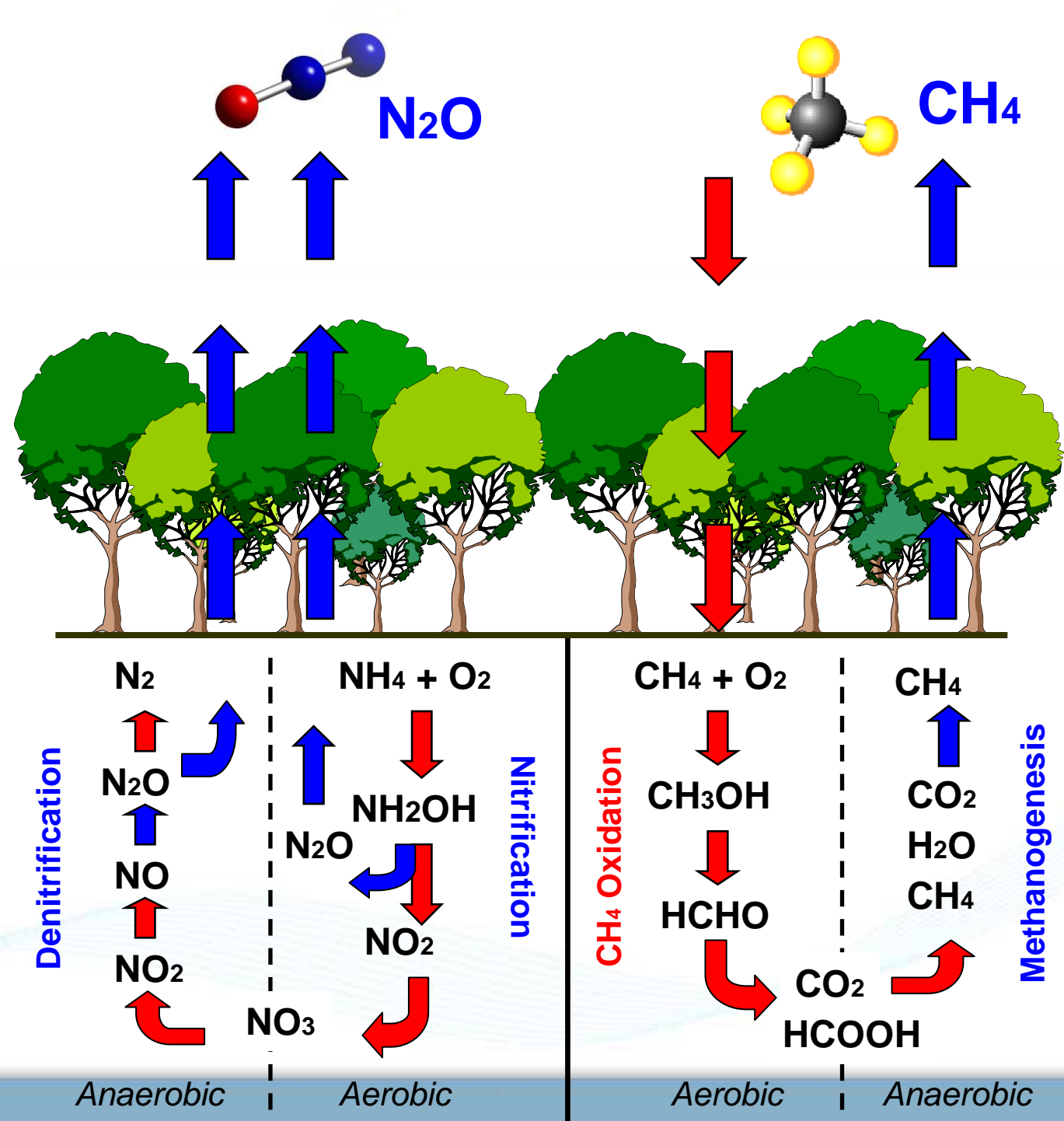
• CO₂



CH₄



N₂O



Measuring soil GHG exchange

Chambers

Manual

- low temporal
- high spatial
- low flux
- CO₂, N₂O & CH₄



Automated

- high temporal
- low spatial
- low flux
- CO₂, N₂O & CH₄



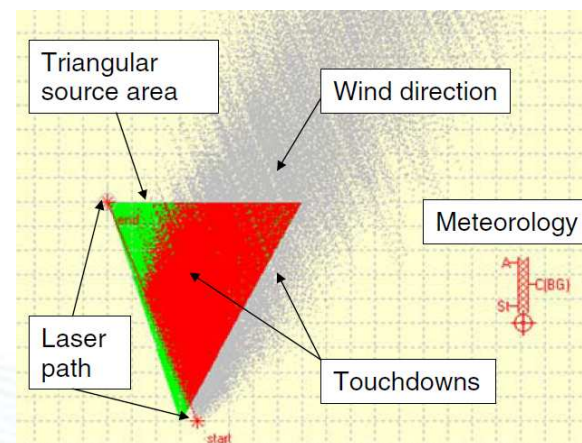
Open

Mass balance

Eddy covariance

Flux gradient

- high temporal resolution
- low spatial
- medium-high fluxes
- CO₂, CH₄ ?, N₂O ??

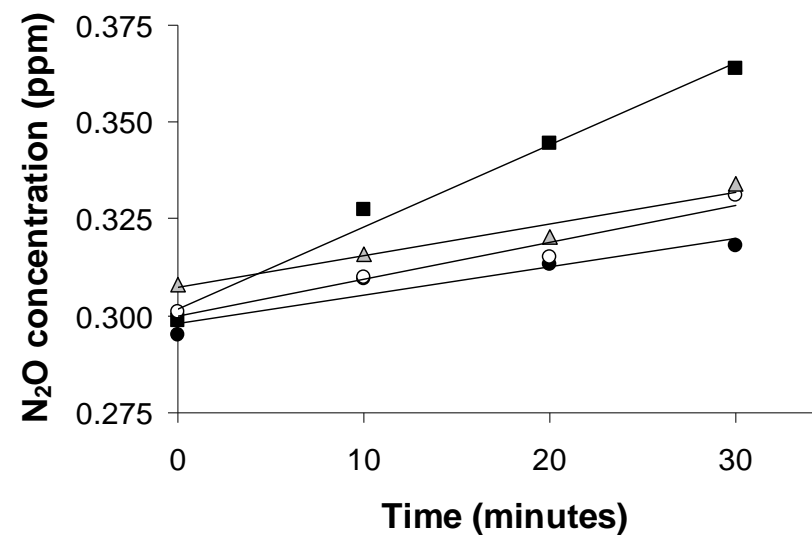
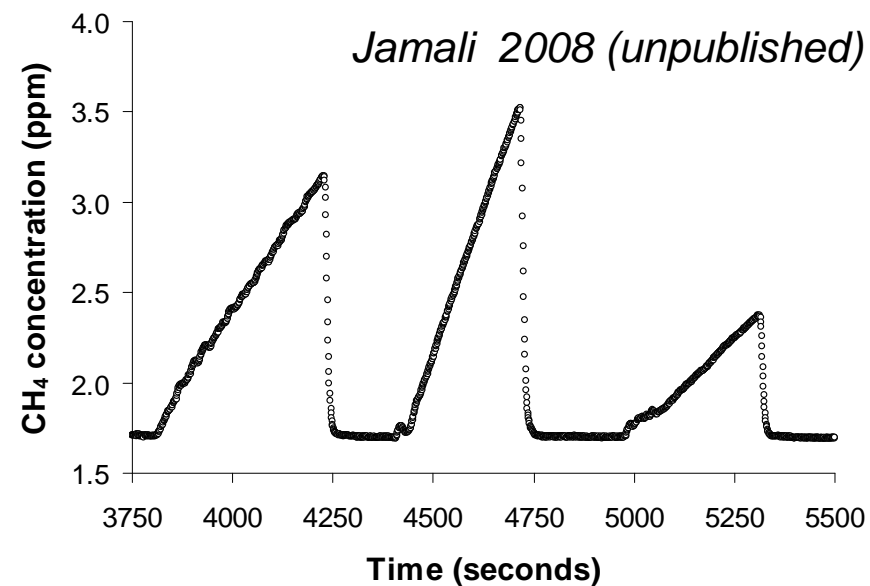
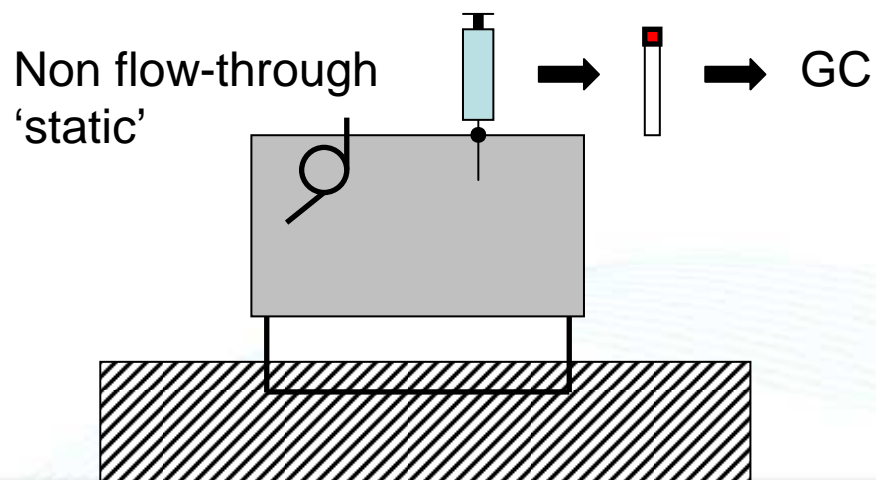
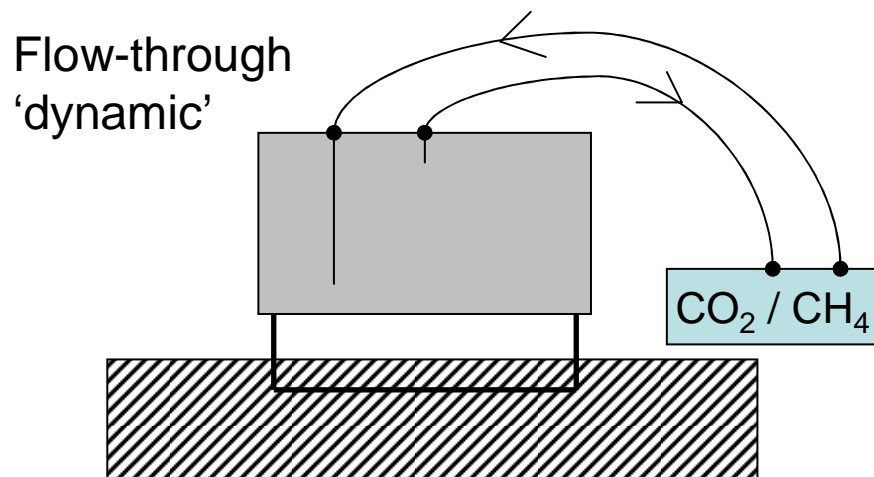


Denmead 2008



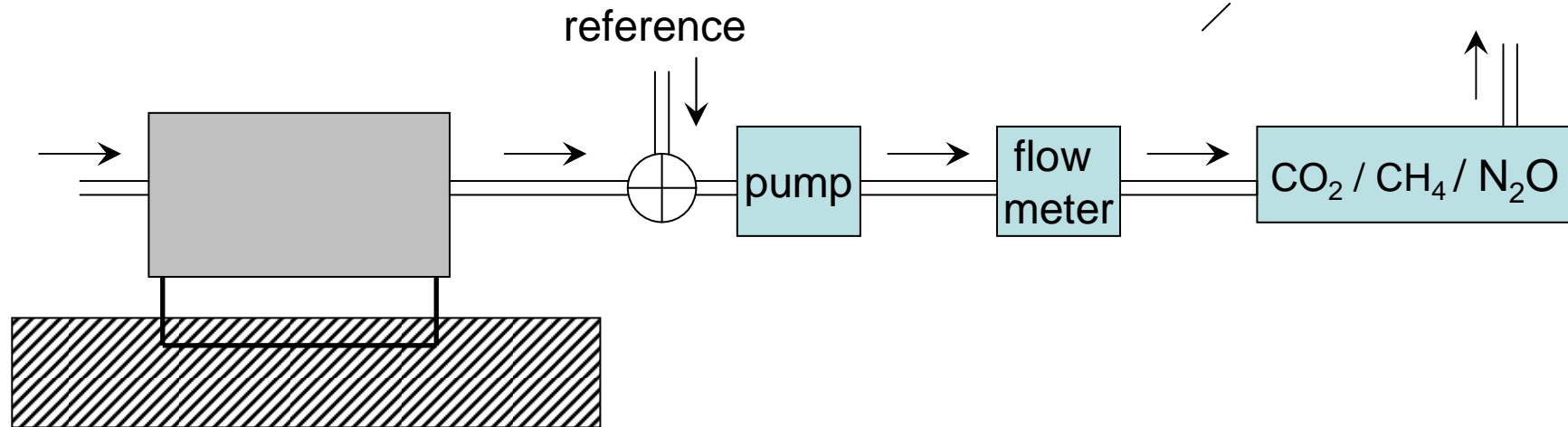
Closed chambers

- Non-steady state (i.e. conc. Increase)



Open chambers

- Steady state (i.e. conc. stable)



- Constant flow of air maintained through the headspace
- The difference in concentration between the air in and out is measured.
- Good – as gas concentration increase minimal and not inhibitory
- Bas – when gas flux is small, therefore difference negligible

Chamber flux calculation

Closed chamber

Parkin et al. 2003

- Flux volume (F_v) as $\mu\text{L m}^{-2} \text{h}^{-1}$ can be calculated from:

$$F_v = (V / A) \Delta_g / \Delta t$$

where V is volume (m^3), A is surface area (m^2), Δ_g is gas concentration change (ppm), and Δt is the period of time (h) for that change.

- Universal gas law converts F_v ($\mu\text{L m}^{-2} \text{h}^{-1}$) to flux density (F_g) $\mu\text{Mol m}^{-2} \text{h}^{-1}$
- Requires data of air temperature and atmos. pressure (altitude).

Open chamber

- Flux density (F_g) as $\mu\text{Mol m}^{-2} \text{h}^{-1}$ can be calculated from:

$$F_g = v (\rho_{g,out} - \rho_{g,in}) / A$$

where v is flow rate ($\text{m}^3 \text{h}^{-1}$) and $\rho_{g,out}$ or $\rho_{g,in}$ are $\mu\text{Mol m}^{-3}$

Denmead 2008

Issues & solutions for chambers

1. Soil Disturbance

(Mosier 1989; Matson & Harris 1995; Denmead 2008)

- *Use temporary/portable chambers.*
- *Install permanent chamber collars at least 24 h prior to flux measure*
- *Collars should be short to minimise above and below-ground perturbation.*

2. Temperature increase:

Can Influence biological activity & physical absorption or dissolution of dissolved gasses.

- *Use insulated, reflective chambers.*
- *Keep deployment time as short as possible.*

3. Pressure perturbations:

Wind may cause pressure-induced mass flow over chamber collar

Closed chamber may reduce natural mass flux.

- *Use vented chamber*
- *Use skirted chambers*

4. Increased humidity

Gas solubility changes but probably a minor effect

Humidity increases may dilute the gas of interest, underestimating the flux

- *Keep chamber deployment short*
- *Relative humidity can be measured to correct for dilution effects from water vapour.*

Issues & solutions for chambers

5. **Diurnal variation** in soil gas flux can follow diurnal temperature, but not universal.
Inter-daily variation may be high due to rainfall, fertility, tillage or freeze thaw events.
Seasonal variation (Spring and Winter fluxes can be substantial).
 - *Measure flux at the time of day that most closely represents daily mean (~ mid morning)*
 - *Temperature correct measured fluxes when temperature bias may occur.*
 - *Measure fluxes 3 to 4 times a week, all year long (**wow !!**)*
 - *Stratify sampling to take account of episodic events (e.g. summer rainfall).*
6. **Spatial variability**
Coefficients of variation commonly exceed 100%.
 - *Use chambers with larger surface area footprint to minimize small scale variability.*
 - *Use as many chambers as possible.*
7. **Gas Mixing**
Assumed molecular diffusion is sufficient to homogenise gas concentrations.
This may not hold with large vegetation present or large volume:surface area.
 - *Pump the syringe repeatedly before sampling from the headspace*
 - *Fit chambers with small fans – a 12 VDC computer fan.*

Minimum detectable limit

Especially when soil fluxes are low, it is important to know the minimum detection limit (MDL).

A function of :

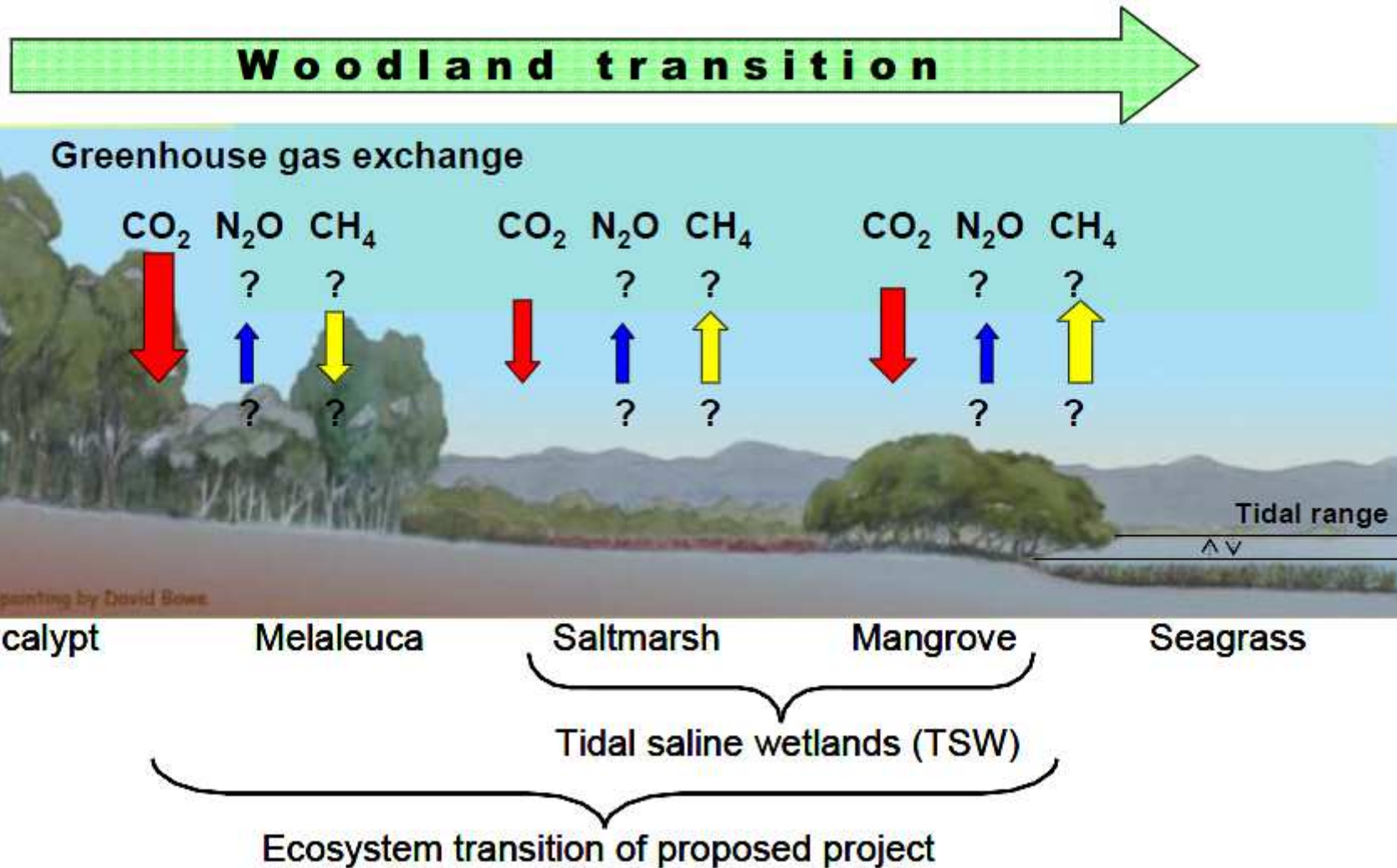
- sampling precision
- analytical precision
- chamber volume and surface area.

Sampling + analytical precision determined from SD of ~ambient standards (n>20).

± 2 SD provides sampling + analytical precision .

$$\text{MDL} = x2 \text{ SD} * \text{volume (L)} / \text{area(m}^2\text{)} / \text{deployment time (min)}$$

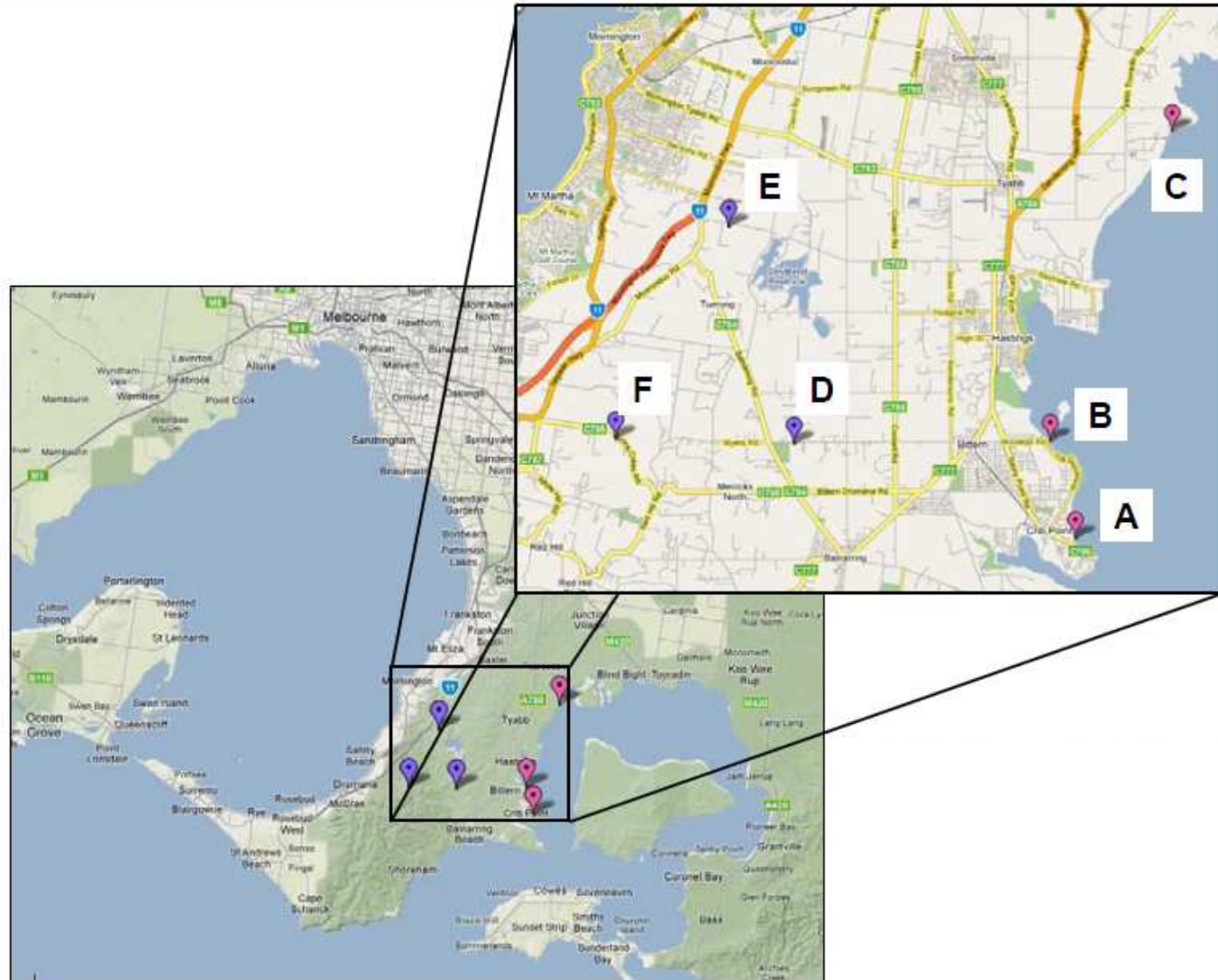
Situations favouring chambers



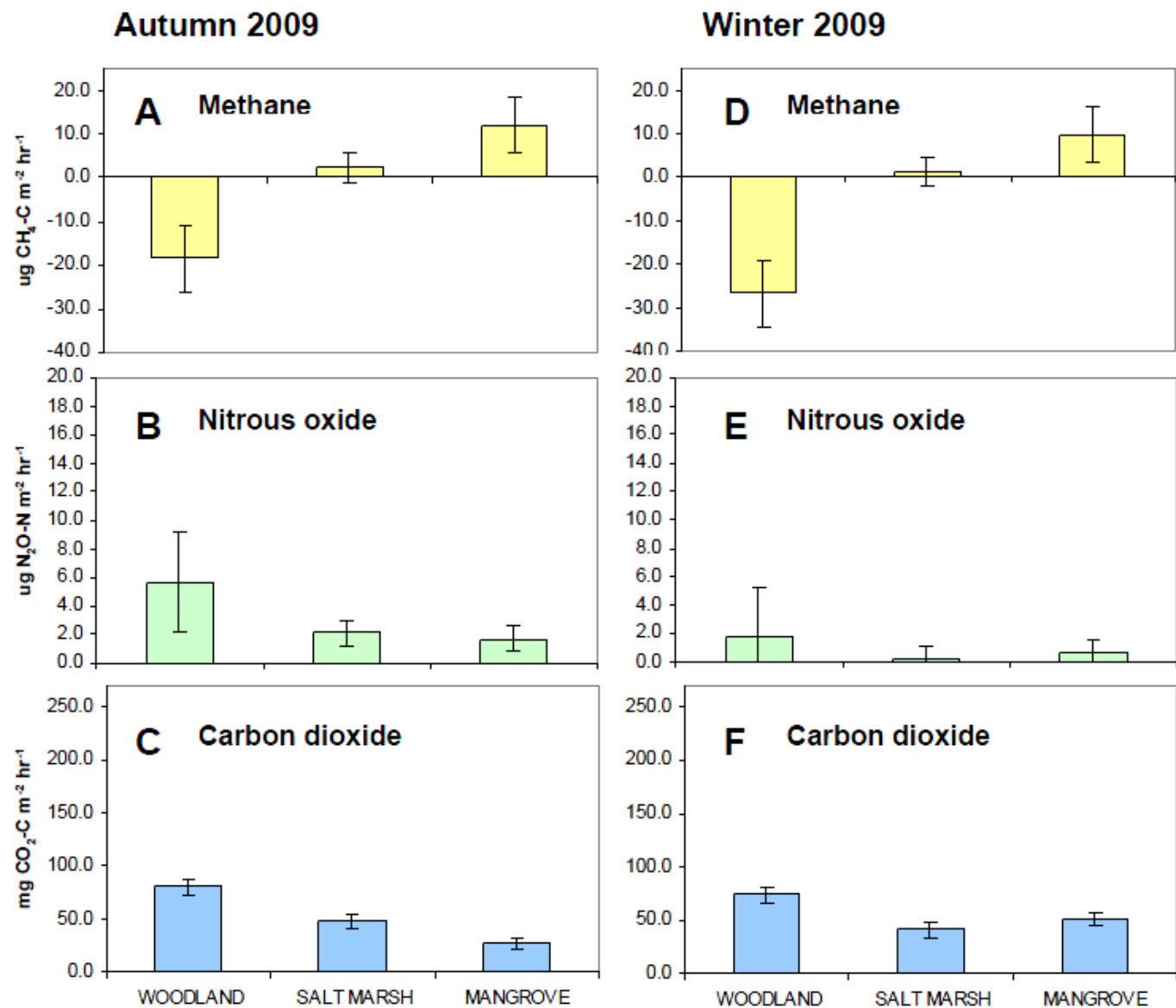


THE UNIVERSITY OF
MELBOURNE

Mornington Peninsula, VIC



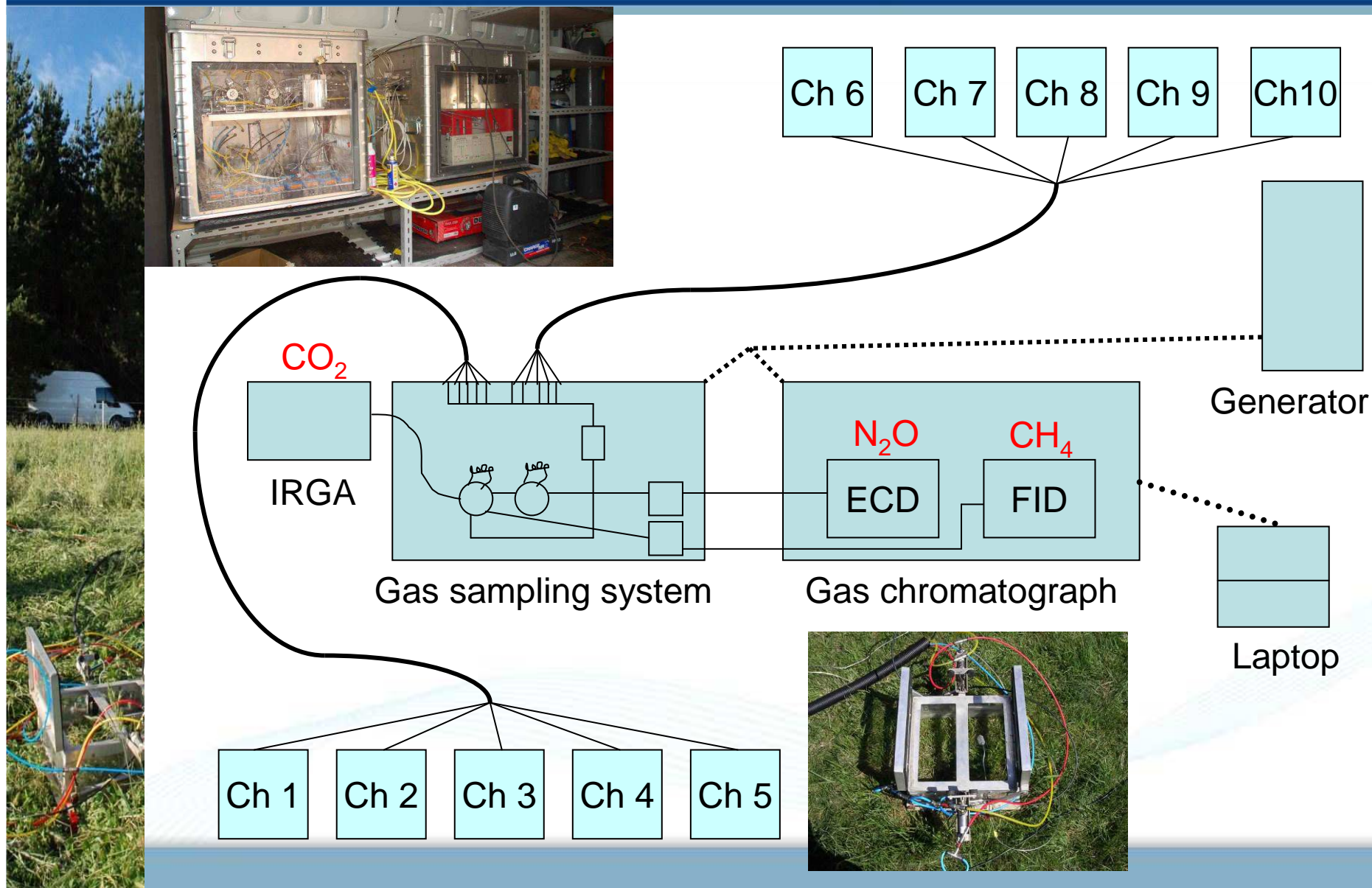
Closed chamber fluxes





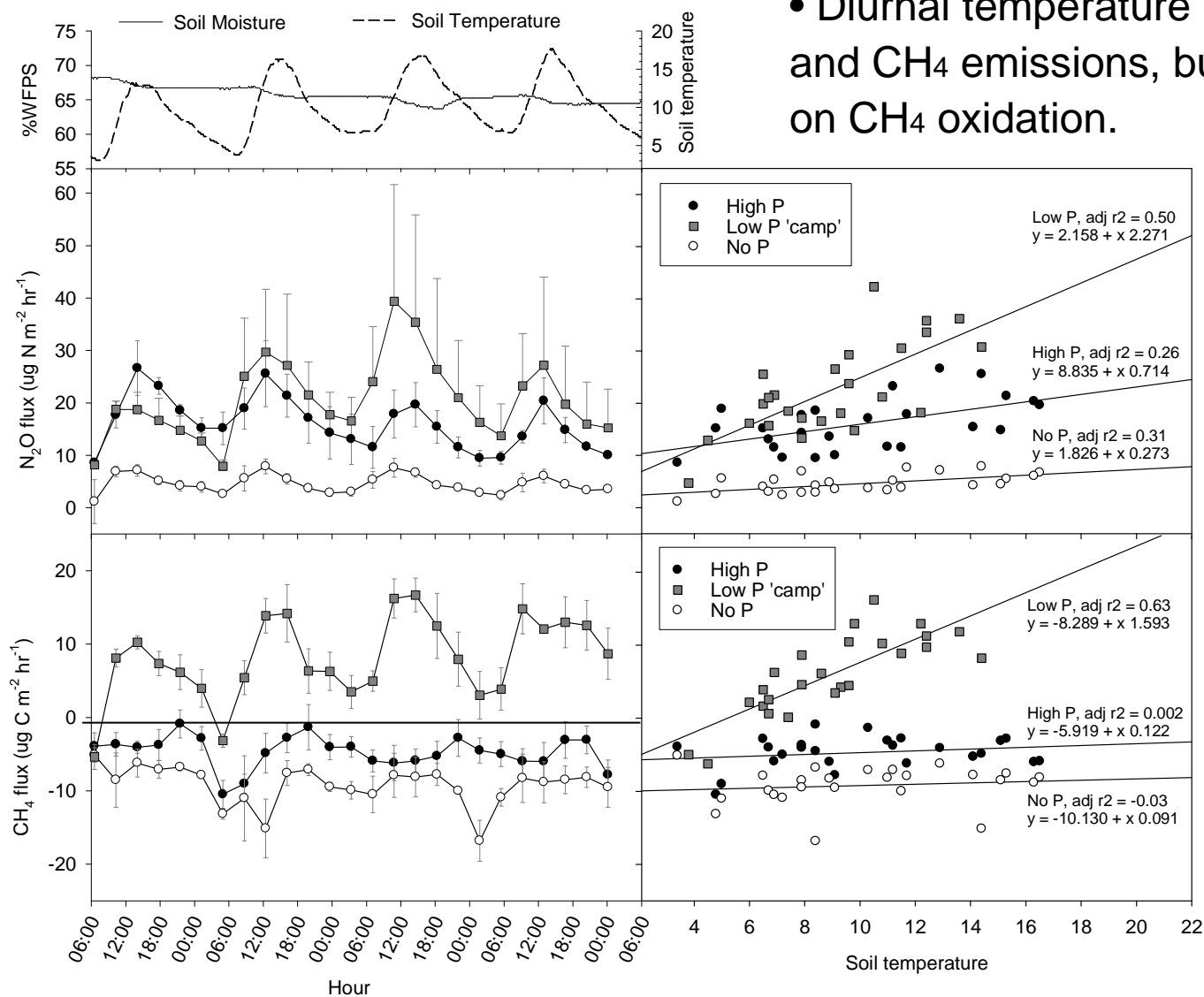
THE UNIVERSITY OF
MELBOURNE

Automated chambers



Diurnal N₂O and CH₄ flux

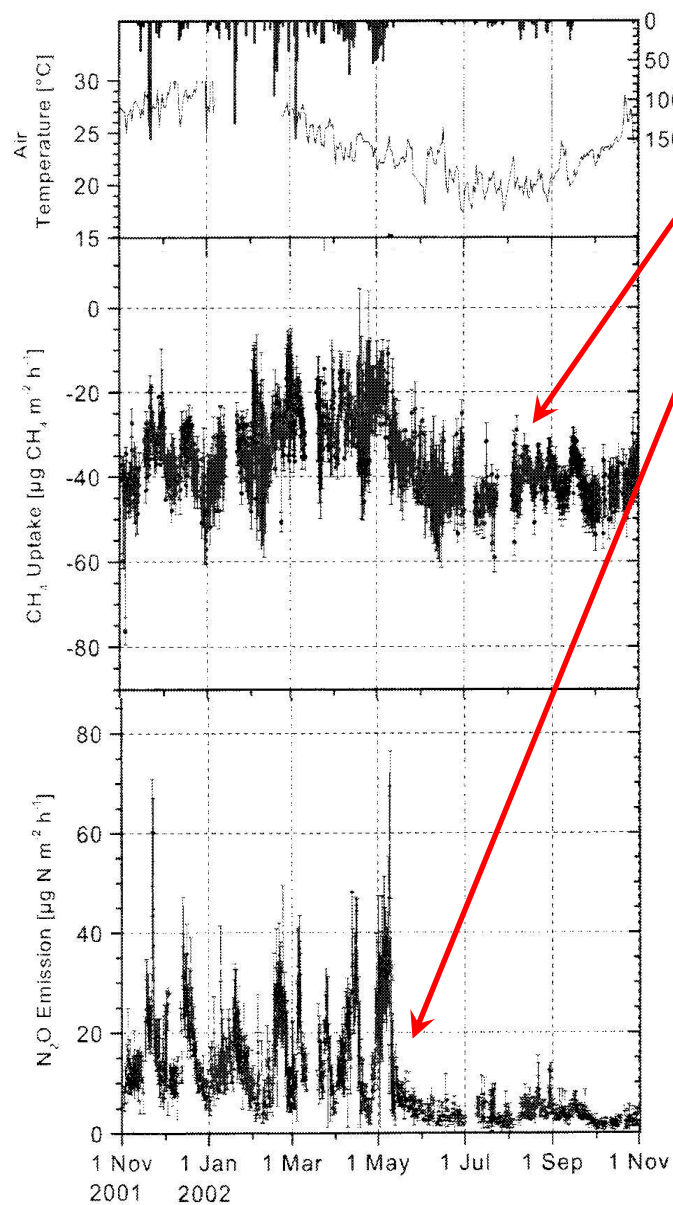
- Diurnal temperature cycle affects N₂O and CH₄ emissions, but has little effect on CH₄ oxidation.



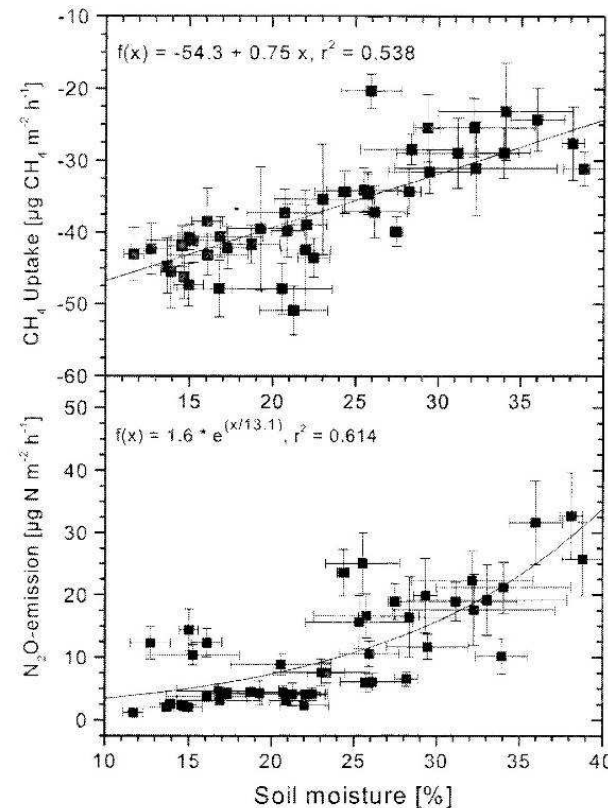


THE UNIVERSITY OF
MELBOURNE

Seasonal N₂O and CH₄ flux



- Seasonal shift in CH₄ uptake rates
- Strong seasonal shift in N₂O emission
- Good correlation between flux and soil moisture.



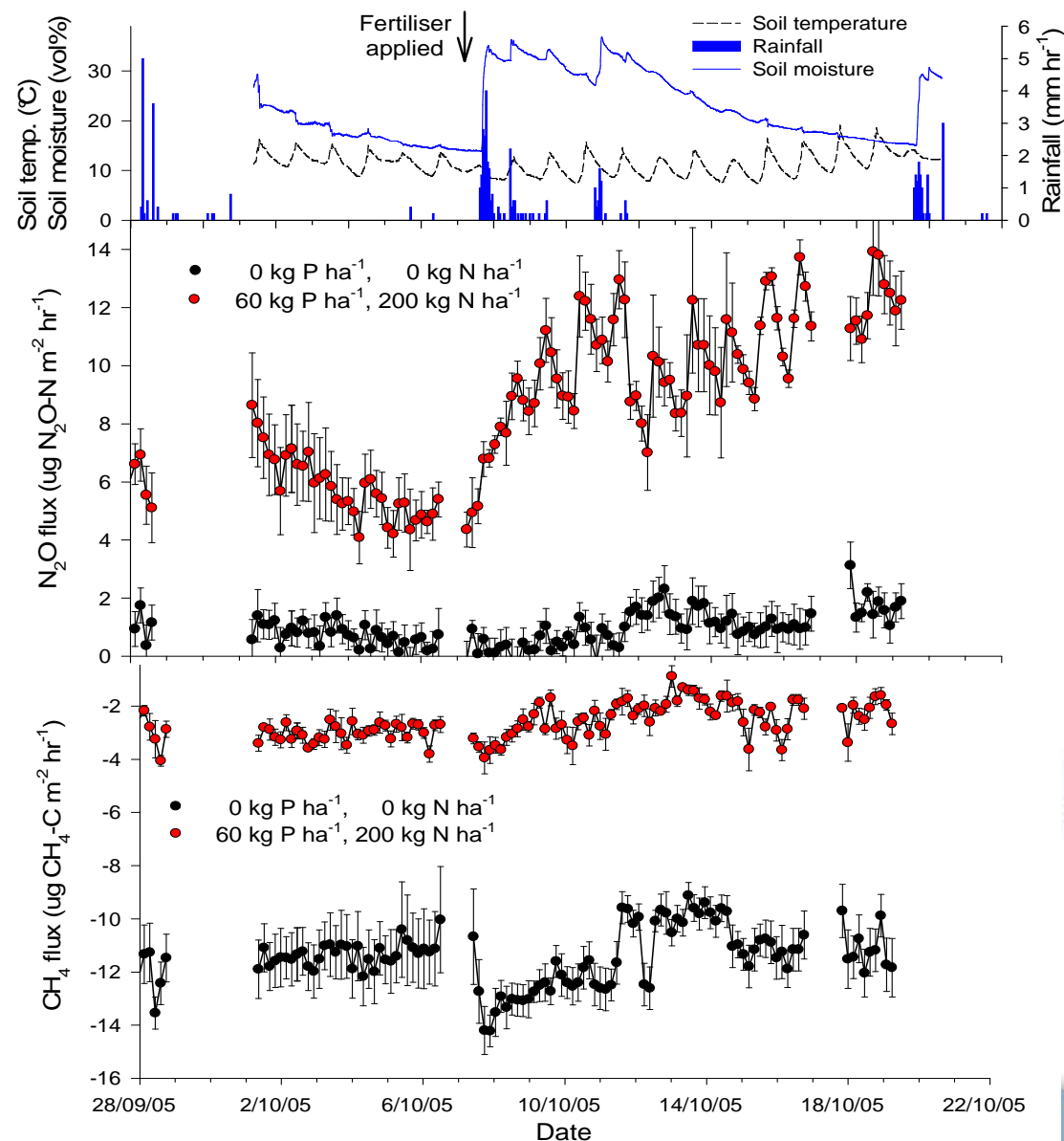
*Kiese et al.,
(2003) & (2005)*

Fertiliser and forest N_2O and CH_4 flux

Annual fertiliser since 1998 @
200 kg N, 60 kg P ha⁻¹

N_2O flux small
($<20 \mu\text{g N m}^{-1} \text{ hr}^{-1}$)
Despite large NO_3 and NH_4
pools, up to 250 mg N kg⁻¹.

X4 less CH_4 uptake with N
fertiliser.



CH₄ and N₂O mitigation opportunities

- There are few opportunities to increase non-CO₂ sinks, except through enhanced CH₄ uptake with agricultural abandonment or active afforestation – *benefit small*
- Mitigation of CH₄ and N₂O flux rests solely at managing the natural and anthropogenic sources.